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# Fluorophosphaalkenes - New Versatile Reagents

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#### FLUOROPHOSPHAALKENES - NEW VERSATILE REAGENTS

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Abstract The synthesis of substituted fluorophosphaalkenes RP=CFR' with a variety of groups R and R' is discussed. The reactivities (self-addition, HX addition, [2+4] cycloaddition with 1,3-dienes) and the coordination chemistry of selected compounds have been explored.

#### INTRODUCTION

Phosphorus carbon  $(p-p)\pi$  compounds show a surprising similarity to alkenes  $^1$ . The proof of this relationship faces some difficulties in case of sterically protected PC double bond systems, in particular for [2+4] cycloaddition reactions  $^2$ . Some successful experiments with heterosubstituted phosphaalkenes  $^3$  demonstrate the influence of substituents, a result which led us to study the chemistry of fluorophosphaalkenes and related compounds.

#### PREPARATIVE ROUTES TO FLUOROPHOSPHAALKENES

 ${
m F_3^{CP=CF}_2}$   $({\scriptstyle \underline{1}})$ , a simple representative of this class of phosphaalkenes was first prepared by Nixon et al. in 1979 by HF elimination from  ${({
m CF}_3)}_2{
m PH}$  and later by Burg by thermal cleavage of the dimer  ${({
m CF}_3^{\ PCF}_2)}_2$ . We recently discovered an optimal method of preparation, using the pyrolysis of  ${
m Me}_3{
m SnP(CF}_3)_2$  according to equ.  ${(1)}^6$ .

$$Me_3SnP(CF_3)_2 \xrightarrow{320°C} Me_3SnF + F_3CP=CF_2$$
 (1)

This method is widely applicable to precursors of the general type  $Me_2SnE(R)R_E$  (E = P, As) and  $Me_2SnE'R_E$  (E' = S, Se) 8.

The preparation of substituted phosphaalkenes RP=CFR' demands suitable pathways to the trimethyltin precursors  $Me_3SnP(R)CF_2R'$ . Their synthesis involves a multistep procedure for the production of the iodophosphanes  $R(R'CF_2)PI$ , from which the required stannyl-phosphanes can be prepared according to equation (2).

The thermal stability of these compounds is strongly influenced by the nature of the substituents R and R'.

Depending on the aim of the actual investigation fluorophosphaalkenes are generated by thermolysis of the corresponding stannylphosphanes either at T >  $300^{\circ}$ C and p  $\approx 10^{-3}$ torr<sup>6</sup> or in solution at 70 - 120°C in the presence of 1,3-dienes as trapping reagents<sup>9</sup>.

Alternatively, heterosubstituted derivatives of  $\frac{1}{2}$  (F<sub>3</sub>CP=CFX) have been prepared by HF elimination from the secondary phosphanes HP(CF<sub>3</sub>)CF<sub>2</sub>X produced by HX addition (X = NR<sub>2</sub>, OR, PMe<sub>2</sub>) to  $\frac{1}{2}^{10}$ . Recent work established a simple one-pot procedure to the C-amino compounds F<sub>3</sub>CP=C(F)NR<sub>2</sub> by the reaction of HP(CF<sub>3</sub>)<sub>2</sub> with HNR<sub>2</sub> in a 1 : 3 molar ratio.

In case of the trimethyltin precursors with two different fluoroalkyl groups  $R_F$  and  $R'_F$  in principle both can participate in the thermal elimination of  $Me_3SnF$ . A careful investigation of the thermolysis of  $Me_3SnP(CF_3)CF_2H$ ,  $Me_3SnP(CF_3)C_2F_5$  and  $Me_3SnP(CF_3)CF(CF_3)_2$  demonstrates that 1,2-elimination preferrably occurs with preservation of the  $CF_3$  and attack of the  $CF_2X$  or  $CF(CF_3)_2$  group. The same tendency is found for the HF elimination from  $F_3CP(H)CF_2X$  with  $NMe_3^{-10}$ . These results can be explained by the so-called "perfluoro effect"  $CF_3$ .

### REACTIVITY STUDIES

The isolation of the new phosphaalkenes in preparative amounts offered the possibility of a systematic investigation of their reactivity. Of particular interest are reactions which can prove or disprove the expected relationship to alkene chemistry.

Self-addition of the fluorophosphaalkenes (T  $\leq$  -78°C) leads to the trans 1,3-diphosphetanes as the main products. However, the cis isomers and the 1,2-diphosphetanes are formed, too, along with small amounts of tri- and polymeric species 6. The self-addition reaction is strongly affected by the substituents on P and C.

In addition to the information recently published on reactions of  $\frac{1}{2}^{10}$  and  $F_5C_2P=C(F)CF_3$  (2)  $^{12}$  with HX compounds we studied (i) the addition of hexamethyldisilazane and (ii) the addition of primary amines RNH $_2$  (R = Bu $^t$ , Pr $^i$ ) to  $\frac{1}{2}$ . In both cases the phosphaalkene undergoes a multistep reaction, involving the intermediates  $F_3CP$  and  $F_3CP(H)CN$  for (i) and  $F_3CP$  and  $F_3CP=C=NR$  for (ii). The final products are the cyclic oligophosphanes ( $F_3CP$ ) $_n$  (n = 4, 5) and Me $_3$ SiF (i) or RNC (ii).

The strong relationship between fluorophosphaalkenes and alkenes has been proved by their pronounced dienophilicity in a series of [2+4] cycloaddition reactions with 1,3-dienes, performed either with the isolated monomers 13 or in a one-pot synthesis with the stannylphosphane precursors 9, 12.

In the area of coordination chemistry our investigations concentrated on two types of reactions:

- (a) The hydrometallation of  $\frac{1}{2}$ , applying main group IV element hydrides  $\operatorname{Me}_3\operatorname{M'H}$  (M' = Si, Ge, Sn) and the hydride complexes  $\operatorname{HM}(\operatorname{CO})_3\operatorname{C}_5\operatorname{H}_5$  (M = Cr, Mo, W). With the exception of  $\operatorname{Me}_3\operatorname{SiH}$  all hydrides react with formation of M'P or MP bonds. In some cases following reactions are observed.
- (b) The ligand properties of  $\frac{1}{2}$  and its derivative  $F_3^{CP=C(F)\,NMe_2}$  ( $\frac{3}{2}$ ) have been studied using  $Cr(CO)_5L$  complexes ( $L=CH_2^{Cl_2}$ , THF) as

precursors. The reaction of  $\frac{1}{2}$  with  $\operatorname{Cr(CO)}_5\operatorname{CH}_2\operatorname{Cl}_2$  at low temperature yields the  $\sigma$ -P complex as the primary product which at room temperature undergoes at least three following reactions: Rearrangement to the  $\pi$ -complex, dimerization to the 1,3-diphosphetane derivative  $[(\operatorname{CO})_5\operatorname{Cr}(F_3\operatorname{CPCF}_2)]_2$ , and formation of a binuclear compound with  $\frac{1}{2}$  acting as a 4e  $(\sigma+\pi)$  bridging ligand.  $\frac{3}{2}$  reacts with  $\operatorname{Cr(CO)}_5\operatorname{THF}$  affording a yellow crystalline complex  $\operatorname{Cr(CO)}_5(F_3\operatorname{CP=CFN})$  Me $_2$ ) for which the coordinating site of the ligand will be eclucidated by an X-ray crystallographic study in the near future.

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